

Introduction

In method development of high-performance liquid chromatography (HPLC), it requires optimization of several conditions, such as bonded-phase, column efficiency, solvent type, pH and temperature. Especially, pH is the most important parameter to control retention, selectivity and sensitivity of ionic compounds in reversed phase HPLC. Although silica based reversed phase columns have been widely used for analytical and preparative separation, they have low stability under alkaline conditions and a limited usable pH range.

Recently, we have developed a new type of organic/inorganic hybrid silica based C18 column, YMC-Triart C18, to improve the chemical stability at expanded pH range and temperature. The novel technologies of manufacturing particles and surface modification provide outstanding chemical stability and excellent peak shape for any kind of compounds under a variety of mobile phase condition.

In this poster, we will show characteristics of this new hybrid C18 column, and some example cases of efficient method development in separation of pharmaceutical compounds and natural products.

Features & benefits of YMC-Triart C18

- Three core technologies for particles and surface modification
 - A multi-layered organic/inorganic hybrid particle
 - A precise granulation with microreactor technology
 - A proprietary C18 bonding and a multi-stage, multi-compound end-capping
- Symmetrical peak shapes and reproducible retention for all types of compounds under a variety of mobile phase conditions
- Improved speed and resolution in UHPLC analysis on 1.9 μm columns with operating pressure up to 100 MPa (14,500 psi)
- Superior column-to-column and lot-to-lot reproducibility provided by YMC's rigorous manufacturing control system
- Outstanding chemical and physical durability over a wide pH range at a high temperature



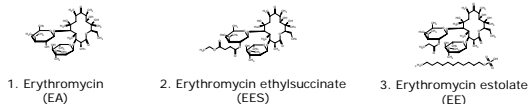
Specification of YMC-Triart C18

Base material	Multi-layered organic/inorganic hybrid
Stationary phase	Polymerically bonded C18 group (USP class: L1)
Particle size	1.9 μm (New), 3 μm, 5 μm
Pore size	120 Å
Carbon loading	Approx. 20%
End-capping	Yes (“multi-stage end-capping” technology)
pH range	1-12
Temperature limit (Recommendation)	70 for pH 1-7 50 for pH 7-12

The advantages of pH and temperature as tools for optimizing resolution and increasing sensitivity

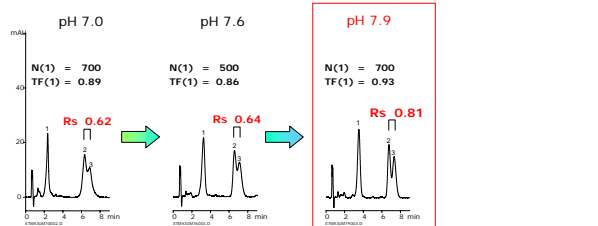
Method optimization of erythromycin and its derivatives by pH and temperature

Structures of erythromycins



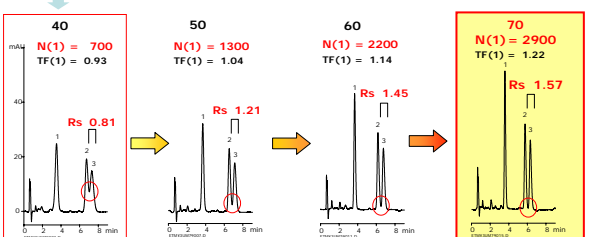
- Erythromycin and its derivatives are macrolide antibiotics which have a broad antimicrobial spectrum.
- Erythromycins are shown to be easily degraded under acidic (< pH 6.5) or strongly alkaline condition. This instability limits the choice of mobile phase condition.

Effect of pH change on separation



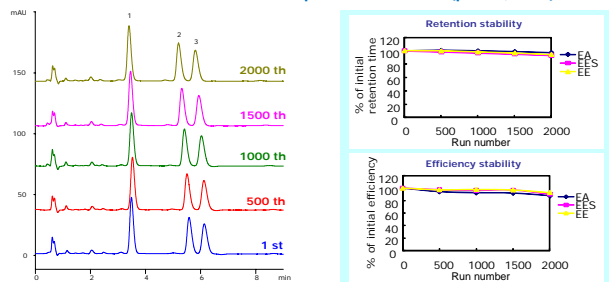
- The pH is optimized within the range in which the compounds are stable.
- The peak shapes and the resolutions are slightly improved by raising the pH.

Effect of temperature change on separation



- The column temperature is increased at pH 7.9, and the higher temperature provides sharper peaks.
- The baseline separation of EES and EE (peak 2 and 3) is achieved at 70 °C.
- The impact of temperature is stronger than that of pH for separation of erythromycins.

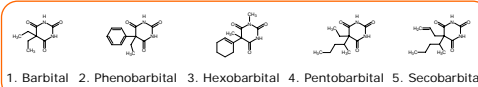
Evaluation of the robustness of the optimized method (pH 7.9, 70 °C)



- The method robustness was evaluated by continuous analysis under the optimized condition.
- No change in retention time and theoretical plate number of erythromycins is observed even after 2000 runs.

Method optimization of barbiturates in human serum by pH and detection wavelength

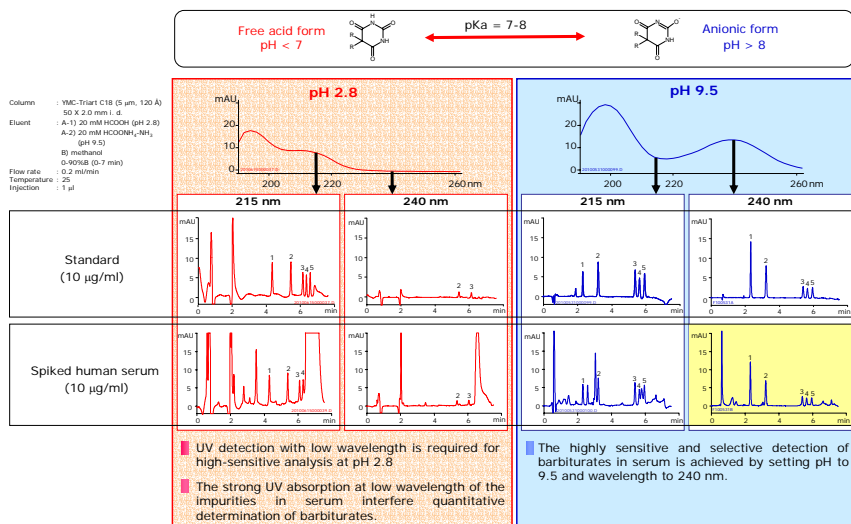
Structures of barbiturates



- Barbiturates have been widely used as psychotropic substances, and some of them are designated to regulate by Convention on Psychotropic Substances.

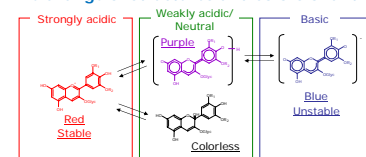
- The structures and UV spectra of barbiturates vary depending on pH. The anionic form at alkaline pH (> 8) has a maximum absorption at 240 nm.

Comparison of separation and sensitivity between low pH and high pH



Analysis of anthocyanins under strongly acidic eluent condition (pH 1.5)

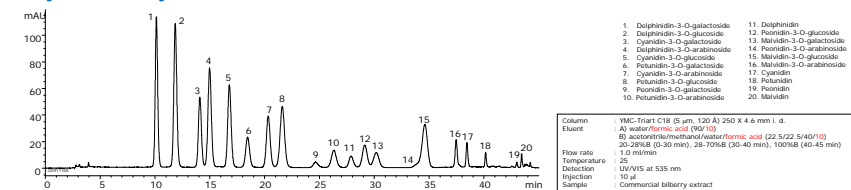
The change of structures and colors of anthocyanins by pH



- Anthocyanin is a kind of flavonoid pigment found in a various fruits and vegetables.

- The chemical structures and the colors of anthocyanins vary depending on pH. The strongly acidic condition is required for reproducible and highly sensitive analysis.

Analysis of bilberry extract with Triart C18



- Triart C18 which has high durability under the strongly acidic condition is suitable for the quantitative analysis and quality control of anthocyanins in bilberry extracts and foods.

Conclusions

- The enhanced durability and chromatographic performance of YMC-Triart C18 offers the maximum flexibility in separation conditions across an expanded pH range.
- 1.9 μm YMC-Triart C18 with excellent chromatographic performance and 100 MPa of maximum operating pressure enables ultra-fast and reliable analysis.
- Identical chromatographic performance and selectivity of Triart C18 across different particle sizes provides mutual method transfer among UHPLC, HPLC and even semi-preparative LC.